## Regioselective Synthesis of Multi-substituted Phenylselenomethyl Isoxazolines

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**Abstract:** A mild, regioselective 1, 3-dipolar cycloaddition protocol for the preparation of phenylselenomethyl isoxazolines through substituted allyl phenyl selenides and nitrile oxides was reported.

Keywords: 1, 3-Dipolar cycloaddition, isoxazolines, selenides.

1,3-Dipolar cycloaddition reactions are very important in the construction of five-membered heterocycles<sup>1</sup>. Nitrile oxides are effective 1, 3-dipoles and can undergo reactions with substituted alkynes and alkenes smoothly to give substituted isoxazoles and isoxazolines, respectively. Both classes of heterocycles are versatile intermediates for the synthesis of natural products and biologically active compounds<sup>2</sup>. But to date, no 1,3-dipolar cycloaddition on selenium-substituted alkynes and alkenes has been reported.

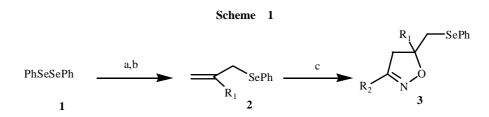
Diorganic selenides have attracted considerable interest because of their potential use as anticancer and antioxidant agents<sup>3</sup>. Our research group<sup>4</sup> has been interested in the application of selenium in organic synthesis for several years. Here we reported a mild protocol to prepare multi-substituted phenylselenomethyl isoxazolines regioselectively.  $\beta$ -Substituted allyl phenyl selenide **2** (1.0 mmol), easily prepared from diphenyl diselenide and  $\beta$ -substituted allyl bromide, was mixed with hydroximoyl halide (1.0 mmol). To the mixture,  $Et_3N$  (1.1 mmol in 5 mL CH<sub>2</sub>Cl<sub>2</sub>) was slowly added dropwise to afford multi-substituted phenylselenomethyl isoxazolines (**Scheme 1**). It is noteworthy that the reactivity of allyl phenyl selenide and  $\beta$ -phenylallyl phenyl selenide in this reaction are higher than  $\beta$ -methylallyl phenyl selenide. The results are summarized in **Table 1**.

In summary, we report a mild 1,3-dipolar cycloaddition protocol through substituted allyl phenyl selenides and nitrile oxides to prepare phenylselenomethyl isoxazolines which have potential use as anticancer and antioxidant agents.

<sup>1279</sup> 

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Wei Ming XU et al.



Reagents and conditions: (a)  $NaBH_4$ , THF, DMF, rt, 1h; (b) substituted allyl bromide, THF, rt, 2 h; (C)  $R_2C(C1)$ =NOH,  $CH_2Cl_2$ ,  $Et_3N$ , rt, 6 h

Product	$\mathbf{R}_1$	$\mathbf{R}_2$	Yield (%) <sup>a</sup>
3a	Н	$4-CH_3C_6H_4$	85
3b	Н	$4-CH_3OC_6H_4$	88
3c	Н	$4-BrC_6H_4$	83
3d	Н	COOEt	97
3e	CH <sub>3</sub>	$4-CH_3C_6H_4$	61
3f	CH <sub>3</sub>	$C_6H_5$	62
3g	CH <sub>3</sub>	$4-BrC_6H_4$	59
3h	CH <sub>3</sub>	$4-FC_6H_4$	53
3i	C <sub>6</sub> H <sub>5</sub>	$4-ClC_6H_4$	83
3j	$C_6H_5$	COOEt	95

Table 1 Synthesis of multi-substituted phenylselenomethyl isoxazolines

<sup>a</sup> Isolated yield.

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1280